

## RAPID COMMUNICATION

### A NOVEL PATHWAY FOR FORMATION OF THIOL METABOLITES AND CYSTEINE CONJUGATES FROM CYSTEINE CONJUGATE SULPHOXIDES

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**Abstract-** *p*-Bromothiophenol and *S*-(*p*-bromophenyl)-*L*-cysteine were formed enzymatically from *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide in the *in vitro* systems with isolated rat hepatocytes or purified cysteine conjugate  $\beta$ -lyases. Isotope dilution study with non-radiolabelled carrier of each product suggested the initial liberation of the thiol and subsequent formation of the cysteine conjugate. C-S bond cleavage pathway to liberate sulphenic acid and thiol are postulated to play an important role in *in vivo* generation of toxic intermediates and products from cysteine conjugates.

A wide variety of organic compounds including drugs, herbicides, industrial materials, and endogenous compounds have been reported to be subjected to glutathione conjugation and subsequent metabolism (1). Conjugation with glutathione and further metabolism generate various kinds of sulphur-containing metabolites, e.g., mercapturic acid conjugates, cysteine conjugates, mercaptopyruvic acid conjugates, mercaptolactic acid conjugates, mercaptoacetic acid conjugates, sulfoxides of these conjugates, thiols, methyl sulphides, methyl sulfoxides, and methyl sulphones (2). In most cases, these biotransformations result in lowered biological activity and rapid elimination of the parent compounds. In the last decade, however, glutathione conjugation and related metabolic pathways have been shown to play a key role in the bioactivation of some nephrotoxic and mutagenic halogenated hydrocarbons (3-6). The implication of cysteine conjugate  $\beta$ -lyases are demonstrated in the biotransformation of cysteine conjugates to the penultimate toxicants, thiols, which are further converted to thionoacetyl fluorides and thioketenes as ultimate reactive intermediates (7-9).

Although the bioactivation pathway via thiols has extensively been investigated, contribution of other pathways to metabolic activation of cysteine conjugates remains to be elucidated. Among such pathways, sulfoxidation of cysteine conjugates and subsequent C-S cleavage to form reactive sulphenic acids have recently been postulated to activate cysteine conjugates (10-12). In the present study, the authors describe the first evidence for *in vitro* liberation of thiols from cysteine conjugate sulfoxides and subsequent transformation of the thiols into cysteine conjugates by the action of cysteine conjugate  $\beta$ -lyase, and discuss possible contribution of the pathway to the generation of toxic intermediates and products from the cysteine conjugates.

## MATERIALS AND METHODS

**Chemicals** - Both [ $^{35}\text{S}$ ]-labelled and non-radiolabelled *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide were prepared from the corresponding *S*-(*p*-bromophenyl)-*L*-cysteine by the method described by Oae et al. (13). *S*-(*p*-Bromophenyl)-*L*-cysteine was synthesized by the method of Saunders (14). Other reagents were commercially obtained. Cysteine conjugate  $\beta$ -lyase [EC 4.1.1.13] was purified from soluble fraction of rat liver and kidney by slightly modified methods previously described (15, 16). Each enzyme preparation gave a single band in SDS-polyacrylamide gel electrophoresis.

**Identification of the Products** - Metabolites of *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide, i.e. M1 (retention time, 4.7 min) and M2 (23 min), were separately collected by HPLC under the conditions

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**Abbreviations.** GC, gas chromatography; HPLC, high-performance liquid chromatography; MS, mass spectrometry; SDS, sodium dodecylsulphate.

described in Incubation with Isolated Rat Hepatocytes. M2 was directly analyzed by GC-MS, while M1 was treated serially with trifluoroacetic anhydride and trimethylsilyl diazomethane prior to GC-MS analysis. The analyses were carried out with a gas chromatograph (Model 5890, Hewlett-Packard, USA) coupled to a mass spectrometer (Model 5988A, Hewlett-Packard, USA). A capillary column (DB-17, 0.25 mm x 10 m, J&W Scientific, USA) was used for GC separation. The column temperature was elevated from 50° to 280° at a rate of 30°/min. Helium at a flow rate of 7.5 ml/min was used as carrier gas. The mass spectrometer was operated in the electron impact ionization mode with an ionization energy of 70 eV.

**Incubation with Isolated Rat Hepatocytes** - Hepatocytes were isolated by collagenase perfusion and subsequent isopycnic centrifugation (17). The cells at a final concentration of  $1.0 \times 10^7$  cells/ml were incubated in Krebs-Henseleit bicarbonate buffer (pH 7.4) containing 2% (w/v) bovine serum albumin and *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide (0.5 mM) at 37° for 30 min. Cell viability estimated by trypan blue staining before and after the incubation was 92% and 77%, respectively. Reaction was terminated by freezing the incubation mixture at -20°. After 30 min the medium was thawed and centrifuged, and the supernatant was subjected to HPLC equipped with Capcellpak C<sub>18</sub> SG120 (4.6 x 250 mm, Shiseido, Japan). *p*-Bromothiophenol and *S*-(*p*-bromophenyl)-*L*-cysteine were eluted with methanol/H<sub>2</sub>O/acetic acid (60:40:1, by vol.) at a flow rate of 0.9 ml/min and detected at 254 nm.

**Enzymatic Reaction by Purified Cysteine Conjugate  $\beta$ -Lyase** - The reaction mixture contained in a final volume of 0.1 ml: radiolabelled or non-radiolabelled *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide (0.5 mM) as substrate, reduced glutathione (1 mM), Tris-HCl buffer (100 mM, pH 7.4), and enzyme (30-70  $\mu$ g). The reaction mixture was incubated at 37° for 30 min under anaerobic conditions. The reaction was terminated by an addition of 0.2 ml of methanol. After centrifugation, the supernatant was subjected to HPLC analysis as described above. When radiolabelled substrate was used, the eluate corresponding to each product was collected for measurement of radioactivity.

## RESULTS AND DISCUSSION

The metabolites of *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide incubated with isolated rat hepatocytes were analyzed by HPLC. Metabolites emerging at 4.7 min (M1) and 23 min (M2) were isolated by HPLC (Fig. 1) and their structures were elucidated by GC-MS directly or after derivatization. The derivatized M1 and M2 metabolites showed a peak at 5.3 and 2.5 min, respectively. Characteristic ions were as follows:  $m/z$  385, 387 [ $M^+$ , relative abundance of 12%],

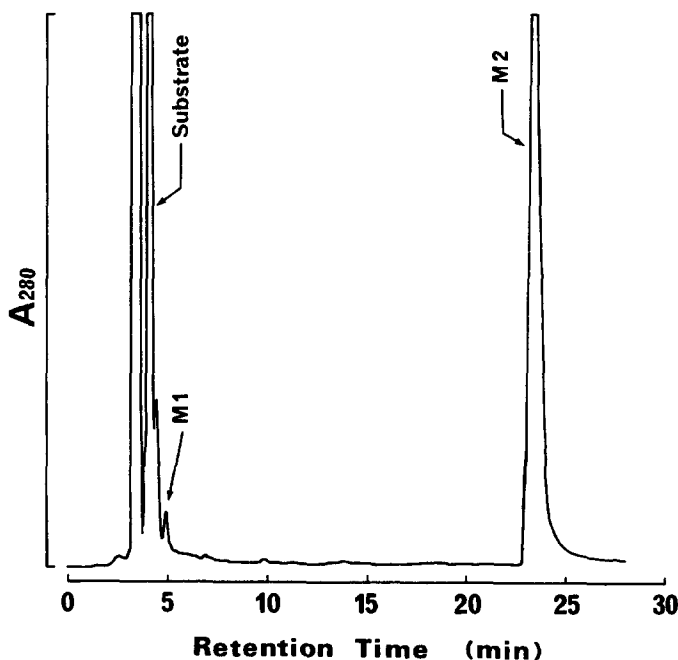
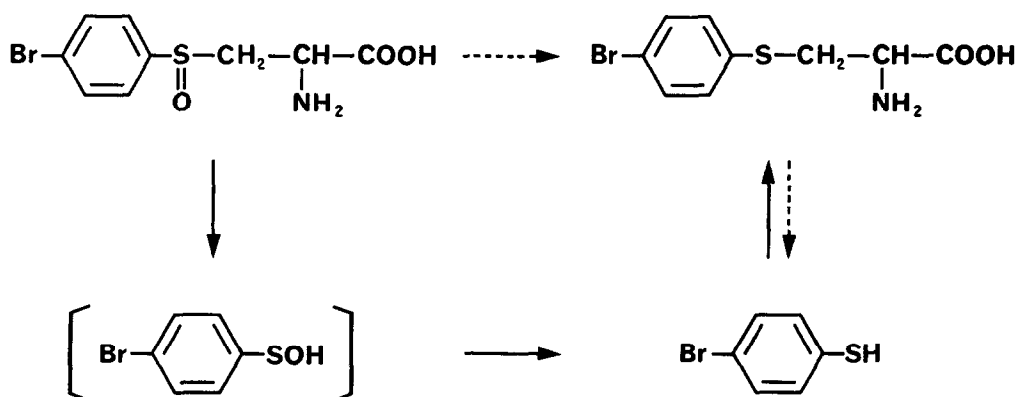


Fig. 1. HPLC profiles of metabolites of *S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide after incubation with isolated rat hepatocytes. *S*-(*p*-Bromophenyl)-*L*-cysteine sulfoxide was incubated with isolated rat hepatocytes and metabolites were analyzed by HPLC. Conditions are described in MATERIALS AND METHODS.



Scheme 1. Possible pathway for formation of thiol and cysteine conjugate from cysteine conjugate sulfoxide

Table 1. Isotope dilution during enzymatic formation of thiol and cysteine conjugate from cysteine conjugate sulfoxide

Additives	Specific radioactivity (μCi/μmole)		
	Substrate	Thiol	Cysteine conjugate
None	1.7 ± 0.3	1.8 ± 0.1	1.8 ± 0.1
Thiol	1.9 ± 0.3	1.1 ± 0.1	0.8 ± 0.0
Cysteine conjugate	1.8 ± 0.1	1.6 ± 0.0	0.6 ± 0.1

[<sup>35</sup>S]-*S*-(*p*-Bromophenyl)-*L*-cysteine sulfoxide was incubated with purified rat hepatic cysteine conjugate β-lyase with or without an addition of non-radiolabelled *p*-bromothiophenol or *S*-(*p*-bromophenyl)-*L*-cysteine as described in MATERIALS AND METHODS. After incubation, the supernatant of centrifuged reaction mixture was subjected to HPLC analysis and radiolabelled products were collected for measurement of radioactivity. Each value represents mean ± S.D. of three experiments.

*m/z* 272, 274 [(*M*-H<sub>2</sub>NCOCF<sub>3</sub>)<sup>+</sup>, 63%], *m/z* 213, 215 [(*M*-H<sub>2</sub>NCOCF<sub>3</sub>-CH<sub>3</sub>COO)<sup>+</sup>, 21%], *m/z* 201, 203 [(C<sub>6</sub>H<sub>4</sub>BrSCH<sub>2</sub>)<sup>+</sup>, 100%], *m/z* 187, 189 [(C<sub>6</sub>H<sub>4</sub>BrS)<sup>+</sup>, 33%] for derivatized M1, and *m/z* 188, 190 [*M*<sup>+</sup>, C<sub>6</sub>H<sub>4</sub>BrSH] for M2. These results were consistent with the proposed structure, i.e. *S*-(*p*-bromophenyl)-*L*-cysteine for M1 and *p*-bromothiophenol for M2. The mass spectra were identical with those of the corresponding authentic compounds. Addition of malononitrile (10 mM) (an inhibitor of pyridoxal enzymes which catalyze β-elimination (18)) to the incubation mixture blocked the formation of M1 and M2 by 100% and 75%, respectively.

Formation of thiol and cysteine conjugate from cysteine conjugate sulfoxide was further confirmed with cysteine conjugate β-lyase purified from rat liver. *S*-(*p*-Bromophenyl)-*L*-cysteine sulfoxide was incubated with the β-lyase in the presence of glutathione. HPLC analysis of the incubation media showed concomitant formation of *p*-bromothiophenol and *S*-(*p*-bromophenyl)-*L*-cysteine. The rate of formation of the thiol and cysteine conjugate was 12 and 1.4 nmole/min/mg, respectively, and was constant during the incubation time until 30 min and with the amount of the enzyme up to 70 μg. Exclusion of glutathione or addition of aminooxyacetic acid (1 mM), a potent inhibitor of cysteine conjugate β-lyase, caused complete inhibition of formation of the two products, although the substrate was consumed even in the absence of glutathione. Similar results were obtained when purified rat renal cysteine conjugate β-lyase was used in place of the hepatic enzyme. These results indicated an involvement of cysteine conjugate β-lyase in the formation of the two products.

Two pathways are conceivable for formation of the thiol and cysteine conjugate (Scheme 1): 1. Initial reduction of the sulfoxide to the cysteine conjugate and subsequent C-S bond cleavage by cysteine conjugate  $\beta$ -lyase to yield the thiol. 2. Initial C-S bond cleavage of the sulfoxide to give a putative intermediate, sulphenic acid, followed by non-enzymatic conversion to the thiol and subsequent transformation of the thiol to the cysteine conjugate by the action of cysteine conjugate  $\beta$ -lyase (reverse reaction). In order to elucidate the operating pathway, we incubated [ $^{35}\text{S}$ ]-*S*-(*p*-bromophenyl)-*L*-cysteine sulfoxide with purified rat hepatic cysteine conjugate  $\beta$ -lyase with or without an addition of non-radiolabelled *p*-bromothiophenol or *S*-(*p*-bromophenyl)-*L*-cysteine (Table 1). Addition of non-radiolabelled thiol decreased the specific radioactivity of the cysteine conjugate formed, while addition of non-radiolabelled *S*-(*p*-bromophenyl)-*L*-cysteine slightly decreased the specific radioactivity of the liberated thiol. In addition, incubation of *S*-phenyl-*L*-cysteine with the  $\beta$ -lyase in the presence of *p*-bromothiophenol yielded *S*-(*p*-bromophenyl)-*L*-cysteine (data not shown). These results indicated that the formation of thiol and cysteine conjugate from the sulfoxide apparently followed the latter pathway and also that the C-S bond formation was dominant in the reversible enzymic reaction between the thiol and the cysteine conjugate.

The intermediacy of sulphenic acid derived from naturally-occurred cysteine conjugate sulfoxide was described in the biotransformation of marasmicine (19) and alliin (20) in plants. The first evidence presented for xenobiotic sulphenic acid formation in mammals was the urinary excretion of 1,1,2,3,4-pentachloro-1,3-butadienyl sulphenic acid in rats dosed with hexachloro-1,3-butadiene (21). Sausen and Elfarrar (12) reported that *S*-(1,2-dichlorovinyl)-*L*-cysteine sulfoxide was much more toxic than *S*-(1,2-dichlorovinyl)-*L*-cysteine *per se in vivo*. In view of the relatively higher chemical reactivity of sulphenic acids compared to the thiols, these findings suggested the conversion of sulphenic acid to thiols may be detoxication pathway. Thus the present study demonstrates a possible alternative pathway for detoxication of sulphenic acids. The highly reactive sulphenic acids may readily be converted by an intermolecular dehydration to give thiosulphinates, which subsequently liberate thiols in the presence of glutathione by thiol-disulphide exchange. The thiols thus generated most probably react with the aminoacylate-enzyme intermediate derived from the enzyme-substrate Schiff base by loss of the  $\alpha$ -proton of cysteine conjugate sulfoxide. Similar reaction mechanism was reported for tryptophan synthase, in which indole reacted with the highly reactive aminoacylate-enzyme intermediate to form tryptophan (22). Together with the lower glutathione concentration in the kidney (11), the essential requirement of glutathione for detoxication of cysteine conjugate sulfoxide found in the present study may explain the kidney specific toxicity of cysteine conjugate sulfoxide. Contribution of the thiol and cysteine conjugate to toxicity as a result of re-involvement in the pathway for sulphenic acid formation still remains to be elucidated.

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